



A new synthetic route to unsymmetric P-chirogenic bisphosphine ligands

Atsushi Ohashi* and Tsuneo Imamoto*

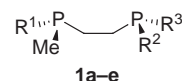
Department of Chemistry, Faculty of Science, Chiba University, Yayoi-cho, Inage-ku, Chiba 263-8522, Japan

Received 29 November 2000

Abstract—A new type of P-chirogenic bisphosphine, unsymmetric BisP*, ligands with different alkyl groups on the two phosphorus atoms was designed and their new synthetic route was developed. A rhodium catalyst with one of the unsymmetric BisP* has demonstrated the ability of highly enantioselective (up to 98%) hydrogenation of the α -dehydroamino acid derivatives under mild conditions. © 2001 Elsevier Science Ltd. All rights reserved.

Optically active phosphine–transition metal complexes have played an important role in catalytic asymmetric synthesis.¹ Among many types of optically active phosphine ligands, the C_2 symmetric bisphosphine ligands have been considered to be endowed with superior properties in catalysis because of their attractiveness augmented by ease of synthesis.² It is well known that C_2 symmetric ligands such as BINAP³ and DuPHOS⁴ have exhibited superior enantioselectivity over a broad front. In a previous report,⁵ (*S,S*)-1,2-bis(alkylmethylphosphino)ethane (alkyl = 1-adamantyl, *t*-butyl, cyclohexyl, cyclopentyl, 1,1-diethyl propyl; abbreviated as BisP*) gave high enantioselectivity in the hydrogenation of various α -dehydroamino acid derivatives when the BisP* was used as a BisP*–Rh catalyst. We had an idea that the enantioselectivity should be enhanced if [(*S*)-1-boranato(R^1)methylphosphino-(*S*)-2-boranato(R^2)methylphosphino]ethane (R^1 , R^2 = 1-adamantyl, *t*-

butyl, cyclohexyl, phenyl; abbreviated as unsymmetric BisP*) with different alkyl groups on the two phosphorus atoms could be used. This means the unsymmetric BisP* has no longer C_2 symmetry.

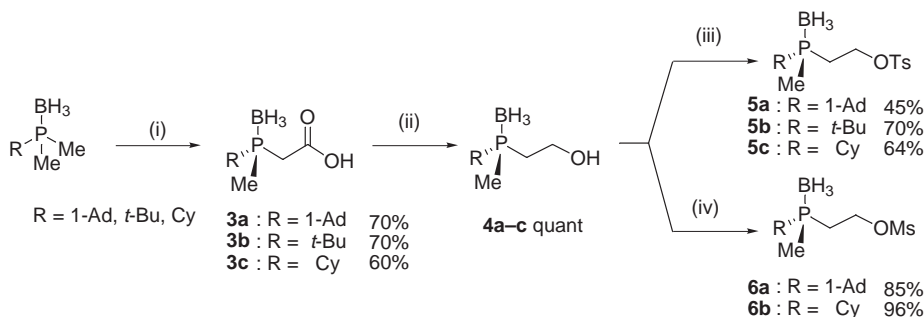


1a–e

Unsymmetric BisP*

- 1a : R^1 = 1-Ad; R^2 = *t*-Bu; R^3 = Me
 1b : R^1 = 1-Ad; R^2 = Cy; R^3 = Me
 1c : R^1 = *t*-Bu; R^2 = Cy; R^3 = Me
 1d : R^1 = 1-Ad; R^2 = R^3 = Cy
 1e : R^1 = 1-Ad; R^2 = R^3 = Ph

This paper reports a new synthetic route of the P-chirogenic bisphosphine, unsymmetric BisP*, (1a–e) and demonstrates the ability of high enantioselectivity



Scheme 1. (i) (a) *s*-BuLi/(–)-sparteine, ether, –78°C, 3 h; (b) CO₂. (ii) BH₃–THF, THF, 0°C, 1 h. (iii) *p*-TsCl, Py, rt, 5 h. (iv) MsCl, CH₂Cl₂, –15°C, 1 h.

* Corresponding authors. E-mail: atsushi@scichem.s.chiba-u.ac.jp

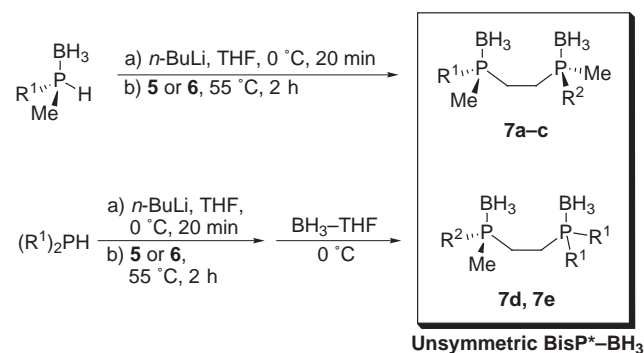
(more than 98%) in the hydrogenation of tetra-substituted α -dehydroamino acids when a BisP*–Rh (**2**) was used as a catalyst.

The synthetic route to the newly designed unsymmetric BisP* is shown in Schemes 1 and 2. Alkyl(dimethyl)phosphine–boranes were enantioselectively deprotonated by treatment with *s*-BuLi in the presence of (–)-sparteine at -78°C ,⁶ and then bubbled with CO_2 to afford (*R*)-1-boranato[alkyl(methyl)phosphino]acetic acids (**3a–c**) (>90% ee) in 60–70% yields (Scheme 1).

Reduction of the carboxylic acids by borane–THF complex afforded in quantitative yields the (*R*)-1-boranato[alkyl(methyl)phosphino]ethanols (**4a–c**), which was treated with TsCl or MsCl to give corresponding (*R*)-1-boranato[alkyl(methyl)phosphino]ethanol 2-tosylates (**5a–c**) or (*R*)-1-boranato[alkyl(methyl)phosphino]ethanol 2-mesylates (**6a–b**), respectively. These compounds were then coupled with (*S,S*)-alkylmethylphosphine–boranes⁷ to provide (*S,S*)-unsymmetric BisP* (**7a–c**)⁸ (>97% ee) together with small amounts of the corresponding diastereoisomers (Scheme 2).

On the other hand, the monochirogenic unsymmetric BisP* (**7d–e**)⁸ was prepared using lithiated dicyclohexyl- or diphenylphosphine as the nucleophiles. The chemical yields of the coupling reactions are shown in Table 1. The combination of 1-adamantyl and *t*-butyl or 1-adamantyl and cyclohexyl groups provided quantitative yield of coupling products **7a** or **7b** (entries 1 and 2), while 1-Ad-BisP* ($\text{R}^1 = \text{R}^2 = 1\text{-Ad}$) was obtained in moderate yield (entry 3). It is noteworthy that this coupling reaction was not effective for the preparation of compound **7c** (entry 4), probably due to the inhibition by undesired by-products. The alternative route to **7a**, in which the positions of R^1 and R^2 were changed, was also successful (entry 5). This procedure was not applicable to the preparation of unsymmetric BisP*, **7d–e** (entries 6 and 7).

An alternative route to **7d–e** was carried out from optically active secondary phosphine–borane with **5d** and **6c** prepared from dicyclohexyl- and diphenyl(methyl)phosphine–borane, respectively (Scheme 3).



Scheme 2.

Table 1.

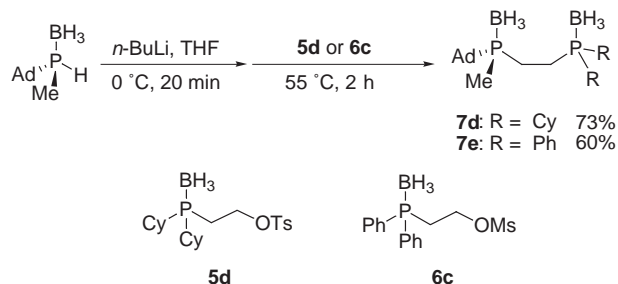
Entry ^a	R ¹	R ²	Yield (%)		
1 ^{b,d}		7a	1-Ad	<i>t</i> -Bu	quant
2 ^d		7b	1-Ad	Cy	quant
3 ^{d,e}		7a'	1-Ad	1-Ad	40
4		7c	<i>t</i> -Bu	Cy	10
5 ^c		7a	<i>t</i> -Bu	1-Ad	quant
6		7d	Cy	1-Ad	trace
7		7e	Ph	1-Ad	35

^aThe mixture of lithiated secondary phosphine–boranes and the mesylates **5** was stirred at 55°C for 2 h unless otherwise noted. ^bThe mixture was stirred at 55°C for 10 min. ^cThe mixture was stirred rt for 1 h. ^dThe tosylates **6** was used. ^eCompound **7a'** is symmetric BisP*.

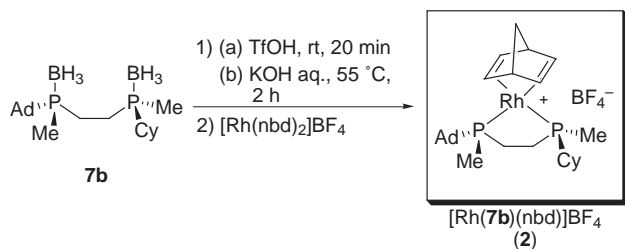
The chemical yields of **7d–e** were improved to 73% and 60%, respectively, probably due to reduced steric hindrance of the nucleophile. One of the unsymmetric BisP*–boranes, **7b**, was converted into the corresponding cationic rhodium complex, **2**, as a quantitative yield according to the reported procedure (Scheme 4).^{8,9}

Preliminary asymmetric catalytic hydrogenation of dehydro-*N*-acetyl- α -cyclohexylglycine methyl ester **8** in the presence of rhodium complex **2** provided the quantitative yield of *N*-acetyl- α -cyclohexylglycine methyl ester **9** with 98.2% ee (*R*). Compared with 1-Ad-BisP* (82.4% ee (*R*)), Cy-BisP* (89.3% ee (*R*)), as a symmetric BisP*,⁵ and (*S,S*)-Me-DuPHOS (96.2% ee (*S*)),¹⁰ this result was drastically improved (Scheme 5).

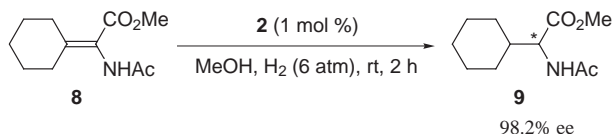
In summary, new P-chirogenic bisphosphine ligands with different alkyl groups on two phosphorus atoms (unsymmetric BisP*) were designed and prepared. In catalytic asymmetric hydrogenation of dehydro-*N*-acetyl- α -cyclohexylglycine methyl ester, the unsymmetric BisP*–Rh gave high reactivity and enantioselectivity. A new class of P-chirogenic bisphosphine, unsymmetric BisP*, will produce a highly asymmetric environment around the rhodium center and will give highly effective enantioselectivity.



Scheme 3.



Scheme 4.



Scheme 5.

Acknowledgements

This work was supported by the Grant-in-Aid from The Ministry of Education, Science, Culture and Sport, Japan.

References

- (a) Noyori, R. *Asymmetric Catalysis in Organic Synthesis*; Wiley & Sons: New York, 1994; (b) Jacobsen, E. N.;

- Pfaltz, A.; Yamamoto, H. *Comprehensive Asymmetric Catalysis*; Springer: Berlin, 1999.
- Kagan, H. B.; Dang, T.-P. *J. Am. Chem. Soc.* **1972**, *94*, 6429.
- Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R. *J. Am. Chem. Soc.* **1980**, *102*, 7932.
- Burk, M. J. *J. Am. Chem. Soc.* **1991**, *113*, 8518.
- Imamoto, T.; Watanabe, J.; Wada, Y.; Masuda, H.; Yamada, H.; Tsuruta, H.; Matsukawa, S.; Yamaguchi, K. *J. Am. Chem. Soc.* **1998**, *120*, 1635.
- Muci, A. R.; Campos, K. R.; Evans, D. A. *J. Am. Chem. Soc.* **1995**, *117*, 9075.
- Nagata, K.; Matsukawa, S.; Imamoto, T. *J. Org. Chem.* **2000**, *65*, 4185.
- Compounds data: **7a**; mp 198–200°C; [α]_D 5.7 (*c* 0.93, CHCl₃); ³¹P NMR (202 MHz, CDCl₃): δ 24.2 (*J*_{PB} = 33 Hz), 28.6–28.9 (*J*_{PB} = 105 Hz). **7b**; mp 197–199°C; [α]_D 2.9 (*c* 1.1, CHCl₃). ³¹P NMR (161 MHz, CDCl₃): δ 19.2 (*J*_{PB} = 49 Hz), 25.2 (*J*_{PB} = 52 Hz). **7c**; mp 136–138°C; [α]_D -12 (*c* 0.29, CHCl₃). ³¹P NMR (202 MHz, CDCl₃): δ 18.2 (*J*_{PB} = 100 Hz), 28.9 (*J*_{PB} = 104 Hz). **7d**; mp 141–143°C; [α]_D -4.0 (*c* 0.71, CHCl₃). ³¹P NMR (202 MHz, CDCl₃): δ 24.2 (*J*_{PB} = 70 Hz), 27.8 (*J*_{PB} = 56 Hz). **7e**; mp 167–169°C; [α]_D 6.4 (*c* 0.67, CHCl₃). ³¹P NMR (202 MHz, CDCl₃): δ 18.2 (*J*_{PB} = 100 Hz), 28.9 (*J*_{PB} = 104 Hz). Complex **2**; ³¹P NMR (161 MHz, CDCl₃): δ 53.3 (*J*_{PRh} = 178 Hz), 57.9 (*J*_{PRh} = 175 Hz).
- (a) McKinstry, L.; Livinghouse, T. *Tetrahedron Lett.* **1994**, *35*, 9319; (b) McKinstry, L.; Livinghouse, T. *Tetrahedron* **1994**, *50*, 6145.
- Burk, M. J.; Gross, M. F.; Martinez, J. P. *J. Am. Chem. Soc.* **1991**, *113*, 8518.